

Amendments to the Specification

Please replace the paragraph at page 2, line 29 through page 3 line 2 with the following amended paragraph:

According to more specific embodiments of the present invention, the first reactant may be a siloxane represented by the general formula  $\text{Si}_n\text{O}_{n-1}\text{X}_{2n+2}$ , where  $n$  is an integer of 2 to 5, and  $\text{X}$  is a chemical group selected from F, Cl, Br, I, or NCO. In a preferred embodiment, the first reactant is a halogen- or NCO- substituted disiloxane (i.e.,  $n=2$ ). Most preferably, the first reactant is a siloxane selected from the group consisting of  $\text{Si}_2\text{OCl}_6$ ,  $\text{Si}_2\text{OBr}_6$ , and  $[\text{or}] \text{Si}_2\text{O}(\text{NCO})_6$  and the second reactant is selected from the group consisting of  $\text{H}_2\text{O}$ ,  $\text{H}_2\text{O}_2$ , ozone ( $\text{O}_3$ ) and  $[\text{or}]$  oxygen radical.

Please replace the paragraph at page 5, lines 19-25 with the following amended paragraph:

The first reactant is generally represented by the formula  $\text{Si}_n\text{O}_{n-1}\text{X}_{2n+2}$ , where  $n$  is an integer of 2 to 5 and  $\text{X}$  is a chemical group selected from F, Cl, Br, I, or NCO. By way of examples, the first reactant as used herein may be selected from the group consisting of  $\text{Si}_2\text{OCl}_6$ ,  $\text{Si}_3\text{O}_2\text{Cl}_8$ ,  $\text{Si}_4\text{O}_3\text{Cl}_{10}$ ,  $\text{Si}_2\text{OBr}_6$ ,  $\text{Si}_3\text{O}_2\text{Br}_8$ ,  $\text{Si}_4\text{O}_3\text{Br}_{10}$ ,  $\text{Si}_2\text{O}(\text{NCO})_6$  and  $[\text{or}] \text{Si}_3\text{O}_2(\text{NCO})_8$ , or mixtures thereof. In a preferred embodiment, the first reactant is a halogen- or NCO- substituted disiloxane. Most preferably, the first reactant is selected from the group consisting of  $\text{Si}_2\text{OCl}_6$ ,  $\text{Si}_2\text{OBr}_6$  and  $[\text{or}] \text{Si}_2\text{O}(\text{NCO})_6$ .

Please replace the paragraph at page 5, lines 27-31 with the following amended paragraph:

The first basic catalyst as used herein is preferably selected from pyridine ( $\text{C}_2\text{H}_5\text{N}$ ) and  $[\text{or}]$  an amine. More preferably, the first basic catalyst is a tertiary aliphatic amine compound having the general formula  $\text{NR}_3$ , where each  $\text{R}$  represents the same or a different aliphatic group having from 1 to 5 carbon atoms. In a specific preferred embodiment, the first basic catalyst is trimethylamine ( $\text{C}_3\text{H}_9\text{N}$ ).

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Please replace the paragraph at page 6, lines 24-26 with the following amended paragraph:

In a preferred embodiment of the invention, the second reactant is selected from the group consisting of  $\text{H}_2\text{O}$ ,  $\text{H}_2\text{O}_2$ , ozone ( $\text{O}_3$ ) and [[or]] oxygen radical. The second basic catalyst may be the same as or different than the first basic catalyst.